

Atomic Structure Electron Configuration Answer Key

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ALLEN

CHEMISTRY TEST PAPER WITH ANSWER.
SECTION-A . 1. The incorrect statement about the imperfections in The valence shell electron .

configuration of the element, which is just above 'E' in the group is (A) 3s. 2. 3p⁴ 10 2(B) 3d . 4s , 4p⁴ ... The correct structure of product 'A' formed in the following reaction. 2 NaOD in D O PhCHO ...

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CHEMISTRY TEST PAPER WITH ANSWER.
SECTION-A . 1. Compound A contains
8.7% Hydrogen, 74% Carbon and 17.3%
Nitrogen. The molecular ... In the
structure of SF₄, the lone pair of
electrons on S is in. ... electronic
configuration (D) atomic mass .
Official Ans. by NTA (D) Allen Ans.
(D) 9. Among the following basic
oxide is : (A) SO₃

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Mar 09, 2003 . 8. When the electron
gains a specific amount of energy, it
moves to a higher orbital and is in
the "excited state". You can
recognize an excited state electro n
configuration. If the configuration
does not match that on the Periodic
Table for that number of electrons,
then it is an excited state.

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B The outer electron in a Sr atom
experiences greater shielding than
the outer electron in a K atom. C The
outer electron in a Sr atom
experiences spin-pair repulsion. D
The outer electron in a Sr atom is
further from the nucleus than the
outer electron in a K atom. 2 What is
the electronic configuration of Mg²⁺?
A 1s² 2s² 2p⁶ B 1s² 2s² 2p⁶ 3s²

**Mark scheme H032/01 Breadth in
chemistry November 2020**

Functional Skills, Key Skills, Entry
Level qualifications, NVQs and
vocational qualifications in areas
such as IT, business, languages,
teaching/training, administration and
secretarial skills. ... 'electron
configuration' for electrons (c) (i)
FIRST CHECK ANSWER ON THE ANSWER LINE
... Gives same answer in powers of 10

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of the periodic table. The valence shell electron configuration of the element, which is just above 'E' in

the group is (A) $3s^2 3p^4$ (B) $3d^1 4s^2 4p^4$ (C) $4d^1 5s^2 5p^4$ (D) $2s^2 2p^4$ Official Ans. by NTA (A) Allen Ans. (A) Sol. E [Ar] $3d^{10} 4s^2 4p^4$ [Ne] $3s^2 3p^4$ 4. Given are two statements one is labelled as . Assertion A and other is ...